**CHE1031 FLIP 3.1: ‘The mole[[1]](#footnote-1)**

**VIDEO:**

‘The mole’ by Bozeman Science, Paul Andersen

https://www.youtube.com/watch?v=Pft2CASl0M0

**Please watch the video and then answer these questions:**

1. Andersen begins with one of his favorite demonstrations. Describe that demo.
2. According to Andersen, the numbers in front of molecules in a balanced chemical equation represent \_\_\_\_\_\_\_\_\_\_\_\_\_?

1. The mole forms a bridge between \_\_\_\_\_\_\_\_\_\_\_ and \_\_\_\_\_\_\_\_\_\_\_\_\_\_.

1. Using Avogadro’s number you can convert grams of water to \_\_\_\_\_\_\_\_\_\_\_ of water and grams of pure iron to \_\_\_\_\_\_\_\_\_\_\_\_ of pure iron.
*Fill in the missing units.*

1. ‘The mole’ by Bozeman Science, Paul Andersen

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